

## The crystal structure of viitaniemiite

AARNE PAJUNEN

*The University of Helsinki*

*Department of Inorganic Chemistry  
Vuorikatu 20, SF-00100 Helsinki 10, Finland*

AND SEppo I. LAHTI

*Geological Survey of Finland  
Kivimiehetie 1, SF-02150 ESPOO 15, Finland*

### Abstract

The crystal structure of viitaniemiite  $\text{Na}(\text{Ca},\text{Mn})\text{AlPO}_4\text{F}_2\text{OH}$   $a = 5.457(2)$ ,  $b = 7.151(2)$ ,  $c = 6.836(2)\text{\AA}$ ,  $\beta = 109.36(3)^\circ$ ,  $V = 251.68\text{\AA}^3$ ,  $Z = 2$ , space group  $P2_1/m$ , has been solved by Patterson and Fourier methods and refined by the least-squares method to an  $R$  index of 0.037 for 728 observed ( $>2\sigma$ ) reflections. The structure contains two sets of infinite chains parallel to the  $b$ -axis, one composed of  $\text{AlO}_2(\text{OH})_2\text{F}_2$  octahedra sharing opposite OH corners and the other of  $(\text{Ca},\text{Mn})\text{O}_4\text{F}_2$  octahedra sharing opposite O-O edges. These chains alternate laterally sharing F corners to form a set of parallel sheets held together by  $\text{PO}_4$  tetrahedra and  $\text{NaO}_4\text{F}_4$  double disphenoids.

The sheet structure of viitaniemiite containing octahedrally coordinated atoms in two separate positions resembles that of montebrasite and eosporite. These three related phosphate minerals are associated with each other in the type locality of viitaniemiite, Viitaniemi pegmatite, Orivesi, southern Finland, where they crystallized during hydrothermal replacement processes caused by residual fluids of the pegmatite melt.

### Introduction

Viitaniemiite occurs as a rare hydrothermal mineral in the phosphate-rich Viitaniemi pegmatite, Orivesi, southern Finland. One of the authors (SIL) has described it as a new mineral in a study on the mineralogy and petrology of the granitic pegmatites of the Eräjärvi area (Lahti, 1981). The mineral was encountered as an inclusion in eosporite aggregate and is associated with morinite, another aluminum-bearing phosphate mineral. Structure analysis confirmed the ideal formula  $\text{Na}(\text{Ca},\text{Mn})\text{AlPO}_4\text{F}_2\text{OH}$  with  $Z = 2$ , although wet chemical analysis indicates that some of the fluorine may be replaced by OH groups. The crystal data measured during the structure analysis are given in Table 1. The X-ray powder data, the optical properties, the chemical data, and the mineralogical description of the mineral have been given by Lahti (1981) and are not reproduced here. A preliminary description of the structure has been given by Lahti and Pajunen (1982). To date viitaniemiite has been identified positively in only three localities: at Viitaniemi, in museum specimens collected from druses of granite in Greifenstein, Sachsen (East Germany), and from Francon quarry, northeastern Montreal, where its occurrence has been confirmed by Ramik et al. (1983). In this quarry viitaniemiite was

encountered as very small crystals in vesicles of silicocarbonatite together with cryolite, calcite, quartz, and weberite. Because the powder diffraction data of viitaniemiite resemble those of an unnamed mineral from Greifenstein, Sachsen, given on JCPDS card 13-0587, the present authors studied several museum specimens taken from this locality and identified viitaniemiite from the samples labeled as lacroixite (cf. Mrose, 1971). According to the descriptions by Slavik (1914, 1915), lacroixite (and therefore also viitaniemiite) occurs there in druses of lithiogranite together with ježekite (sodian morinite), apatite, childrenite (Fe end member of the isomorphous series childrenite-eosporite), roscherite, and tourmaline. Preliminary microanalyzer determinations showed that Greifenstein viitaniemiite is rich in calcium and contains only a few percent of manganese; the fluorine content, however, is equal to that of the Finnish viitaniemiite. Due to its different chemical composition, the Greifenstein viitaniemiite has a larger unit cell ( $a = 5.48\text{\AA}$ ,  $b = 7.18\text{\AA}$ ,  $c = 6.85\text{\AA}$ ,  $\beta = 109.00^\circ$ ,  $V = 254.84\text{\AA}^3$ ; based on the precession films from sample no. 86746 in Harvard Mineralogical Museum). Detailed studies on the Greifenstein viitaniemiite and lacroixite are in progress, because some of the data cited for lacroixite obviously derive from viitaniemiite.

ITANIEMITE NACCA AND ALPO4F2OH											
0,0,L	8	241	255	-6	235	-235	1,3,L	-6	36	-41	-4
1	115	-98	0,5,L	-4	191	-185	-8	66	-62	-5	164
2	667	622		-3	613	606	-7	37	-40	-2	156
3	411	-372	1	145	-146	-2	137	-135	-6	351	-3
4	400	371	2	211	211	-1	271	276	-5	327	202
5	65	-54	3	80	88	0	133	139	-3	127	-1
6	369	361	4	106	107	1	286	272	-2	130	290
7	233	-232	5	116	117	2	561	-537	-1	529	-304
8	321	324	6	84	-84	3	543	524	0	511	721
9	116	-120	7	59	58	4	92	-91	1	453	513
						5	514	505	2	440	440
							241	-238	6	175	214
								-238	46	5	56
									-42	6	49
									7	150	142
										7	-148
0,1,L	0,6,L	6	94	-89	3	161	162				
1	190	-185	0	20	-15	8	50	-53	5	114	120
2	389	382	1	34	32			6	48	53	-6
3	172	157	2	206	199	1,1,L				132	126
4	166	158	3	187	-184	-9	120	120	-3	43	41
5	187	186	4	189	184	-9			-2	44	-43
6	93	-95	5	62	-62	-8	35	33	-1	124	119
7	79	76	6	249	251	-7	55	-53	0	49	-54
8	38	39	7	143	158	-6	202	203	2	86	-5
						-4	23	28	1	127	125
						-4	23	28	2	81	-3
						-4	53	316	3	57	69
						-2	320	-305	5	57	-58
						-3	44	452	5	73	-370
						-2	320	-305	-73	0	43
0,2,L	0,7,L	-3	44	53	-4	316	-293				45
1	219	205	1	49	50	-1	165	166	-2	101	101
2	152	149	2	173	-170	0	54	58	-1	294	276
3	97	-78	3	16	-13	1	289	-303	0	115	-112
4	397	-394	4	42	-45	2	108	112	1	89	141
5	357	349	5	99	-105	3	44	-51	2	221	212
6	29	28	6	66	64	4	42	-33	3	500	-201
7	268	270				5	89	82	4	131	5
8	164	172	0,8,L	6	145	-148	5	298	326	-1	158
	112	118		8	40	39	6	57	-60	0	161
							6	43	-42	2,2,L	217
0,3,L	0	464	453			7	70	75	1	81	20
1	60	-58	1	60	-58	1,2,L			2	80	-17
2	316	319				1,5,L			3	302	144
1	184	-169	3	32	-34	-9	298	300	4	106	146
2	394	-377	4	61	58	-8	114	-113	-8	316	-51
3	81	76	5	59	-65	-7	384	373	-7	106	363
4	41	-36				-6	201	-197	-6	195	-193
5	249	-257	0,9,L	-5	263	248	-4	31	27	822	805
6	28	24		-4	169	166	-3	42	-70	341	-324
7	30	-29	2	129	128	-3	749	671	-2	94	155
			3	49	-45	-1	871	851	-1	74	160
						0	364	-379	1	74	-641
						1	405	432	2	124	-653
						1	212	-208	3	124	217
0,4,L	0,10,L	0	364	-379	0	43	42	2	40	-40	-197
0	1119	1089	2	212	-196	2	82	88	2	431	642
1	88	-75	0	63	63	3	249	258	3	144	-333
2	610	603	4	264	286	5	56	56	5	188	441
3	143	-132	1,0,L	5	347	351	6	128	-124	6	193
4	157	156		6	72	-72			-9	101	321
5	90	-94	-9	165	170	7	191	203	-8	257	-32
6	344	346	-8	185	185	8	110	-116	-7	151	150
7	146	-145	-7	69	61				-6	85	197
						-7	309	297	-5	184	197

ITANICIRITE NAC(MN)ALP04F204

2,3,L		2,6,L		-3	63	65		3,3,L		-2	37	-37	6	233	223
-8	26	-27	-7	51	-46	-6	221	-235	-9	38	-38	0	32	-33	4,1,L
-7	42	41	-6	345	339	-5	306	309	-8	56	-52	1	186	193	
-6	108	-104	-5	91	-85	-4	463	-491	-6	227	-224	2	40	-46	-9
-5	246	247	-4	407	394	-3	307	321	-5	39	38	3	69	70	-8
-4	203	199	-3	249	-244	-2	246	-245	-4	132	138	5	174	174	-7
-3	139	-132	-2	214	205	-1	220	219	-3	147	-148				-6
-2	128	124	-1	138	-137	0	152	155	-2	78	83		3,7,L		-5
-1	94	91	0	67	64	1	540	535	-1	86	84			-4	104
0	31	31	1	172	-183	2	224	-215	0	83	83	-6	95	-94	-3
1	29	35	2	385	402	3	25	244	1	193	201	-5	20	-11	-2
2	336	-349	3	124	-127	4	1	-180	2	126	-128	-4	34	33	-1
3	77	-76	4	282	304	5	241	240	4	123	122	-3	71	-75	0
4	39	37	5	142	-155	6	122	117	5	70	-73	-2	140	137	1
5	155	-153	6	121	130	7	80	72	6	34	28	-1	33	34	2
7	26	-16								0	29		-32	3	92
										1	91		4	38	86
										2	46		5	138	129
2,4,L		2,7,L		3,1,L		3,4,L									
-8	171	165	-6	65	-65	-9	108	110	-7	229	226	3	19	8	
-7	148	-142	-5	84	86	-8	109	115	-6	129	-127	4	56	55	4,2,L
-6	183	176	-4	72	66	-7	56	-56	-5	355	349				
-5	199	188	-3	39	-31	-6	153	157	-4	411	-411		3,8,L		-9
-4	203	192	-2	143	138	-5	34	35	-3	146	145				-8
-3	134	137	-1	18	-16	-3	94	99	-2	125	-120	-5	246	241	-7
-2	134	137	0	65	-60	-2	269	-275	-1	285	284	-4	237	-236	-6
-1	343	334	1	71	70	-1	75	-80	0	110	125	-3	94	92	-5
0	85	-79	2	125	-124	0	65	65	1	306	309	-2	36	-36	-4
1	453	468	3	49	-41	1	206	-214	2	191	-193	-1	181	179	-3
2	153	164	5	112	-113	2	20	21	3	268	279	0	58	62	-2
3	343	358				3	19	-24	4	125	-122	1	156	168	-1
4	237	252		2,8,L		4	78	-70	5	172	174	2	108	-113	0
5	57	60				5	123	117	6	64	66			1	137
5	93	-96	-5	98	94	6	64	-63					3,9,L	2	219
6	116	116	-4	91	83	7	31	-24		3,5,L		-2	47	-44	3
7	98	-101	-3	23	25							-2	27	-27	4
			-2	231	232		3,2,L		-8	104	101	0	27	27	5
2,5,L		-1	57	-58					-7	54	-53				4,3,L
-8	25	23	0	258	263	-9	154	150	-6	102	100		4,0,L		
-7	128	-126	1	25	18	-8	52	-48	-5	33	23				4,3,L
-6	67	67	2	207	212	-7	348	344	-3	79	81	-9	143	-146	-8
-5	66	-62	3	136	141	-6	143	145	-2	198	-199	-8	344	360	-7
-4	72	-67	4	46	48	-5	153	152	-1	20	-19	-7	67	-66	-6
-3	32	-32		2,9,L		-3	326	334	0	61	64	-6	136	139	-5
-2	248	-243				-2	56	-56	1	145	-152	-5	132	-139	-4
-1	55	54	-4	74	-72	-1	592	620	5	109	107	-3	109	107	-2
0	157	156	-3	65	54	0	14	14				-2	753	772	-1
1	85	-94	-2	71	-69	1	423	439		3,6,L		-1	273	-275	0
2	126	141	1	30	-36	2	49	-52				0	442	444	57
3	50	53	2	131	136	4	57	-55	-7	207	202	1	118	-118	56
4	49	47				5	281	278	-6	142	139	2	292	280	91
5	132	134		3,0,L		6	41	-47	-5	188	187	3	39	35	54
6	48	-48							-4	44	47	4	256	248	53
			-9	107	115				-3	133	123	5	99	-91	-96

4,4,L	-3	70	-69	-7	255	258	-2	120	-118	-2	19	-14	0	88	86	
-8	280	278	-1	90	93	-6	74	76	-1	98	-94	-1	70	-67		
-7	126	-128	0	24	7	-5	278	278	0	53	51	0	114	108	6,6,L	
-6	129	129	1	71	-58	-4	97	100	1	77	-81	1	51	-50		
-5	43	-13	1	43	41	-3	405	412				2	55	43	-3	
-4	257	252	2	49	-51	-2	179	-178		5,6,L	3	56	52	-2	72	
-2	538	551		4,8,L	0	143	-138	-6	30	27		6,2,L		7,0,L		
-1	138	-139			1	233	233	-5	244	245						
0	387	397	-4	143	136	2	106	100	-4	94	95	-7	42	-43	-6 196 -193	
1	123	-128	-3	57	-50	3	101	93	-3	234	232	-6	307	307	-5 181 171	
2	180	181	-2	301	302	4	74	-70	-2	176	-174	-5	211	-208	-4 154 -142	
3	53	54	-1	75	-75				-1	252	251	-4	161	161	-3 133 129	
4	234	238	0	238	241			0	64	-63	-3	111	-100	-2 46 47		
5	84	-86	1	76	-86			1	152	149	-2	156	148	-1 114 103		
					-8	60	-61	2	28	29	-1	25	23	0 110 -101		
4,5,L		5,0,L			-7	48	45				0	215	215			
-7	99	-93	-7	139	143	-5	65	-66			1	152	-141	7,1,L		
-4	60	-61	-6	95	-98	-4	140	140	-4	58	56	2	223	211	-6 48 46	
-3	97	96	-5	215	217	-3	42	-42	-3	60	-58		6,3,L	-5 37 34		
-2	141	-146	-4	142	-137	-2	76	75	-2	62	62			-4 87 -80		
-1	27	-29	-3	119	119	-1	61	61	-1	78	77	-7	115	111	-3 43 37	
0	119	120	-1	41	-46	1	166	163	0	28	-27	-6	74	-71	-2 32 -32	
1	37	-38	0	34	-30	3	59	-56				-5	61	-61	-1 116 -107	
2	73	74	1	346	339	4	55	55		6,0,L	-4	92	92	0 25 33		
3	58	57	3	317	292						-3	29	-33			
4	29	-22	4	105	-93			5,4,L	-8	179	183	-1	23	23	7,2,L	
									-7	97	102	0	109	-100		
4,6,L		5,1,L			-7	71	70	-6	145	143	1	70	71	-6 95 92		
-7	66	-62	-8	132	138	-5	258	253	-5	62	59	2	30	-28	-5 72 70	
-6	177	171	-7	30	-26	-4	83	-79	-3	52	-51		6,4,L	-4 66 60		
-5	96	-88	-6	87	88	-3	63	65	-2	321	309			-3 258 250		
-4	229	230	-5	37	41	-2	59	-54	-1	30	23	-6	86	82	0 70 -65	
-3	25	-27	-4	77	-82	0	22	22	0	208	202	-5	68	66		
-2	93	94	-3	104	102	1	279	274	1	85	85	-4	82	80	7,3,L	
-1	90	97	-2	152	-148	2	35	-36	2	26	22	-3	61	-65		
0	24	25	-1	120	-124	3	253	251	3	35	24	-2	229	225	-5 73 -70	
1	141	-144	0	30	38							0	214	206	-4 74 73	
2	131	135	1	118	-119			5,5,L		6,1,L	1	80	74	-2 47 38		
3	40	-41	3	29	25	4	73	-68	-7	29	-23	-8	86	90	-1 82 74	
									-6	62	61	-7	78	-74		
4,7,L		5,2,L			-5	29	26	-6	36	34	-4	59	-53	7,4,L		
-6	41	-39	-8	181	-184	-3	80	80	-3	94	94	-3	79	77	-3 130 120	
-4	87	88										-1	66	-64		